



VISUAL
PHYSICS

SHORT NOTES

C H A P T E R

Heat Transfer

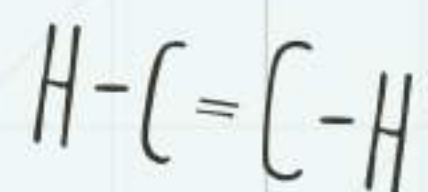
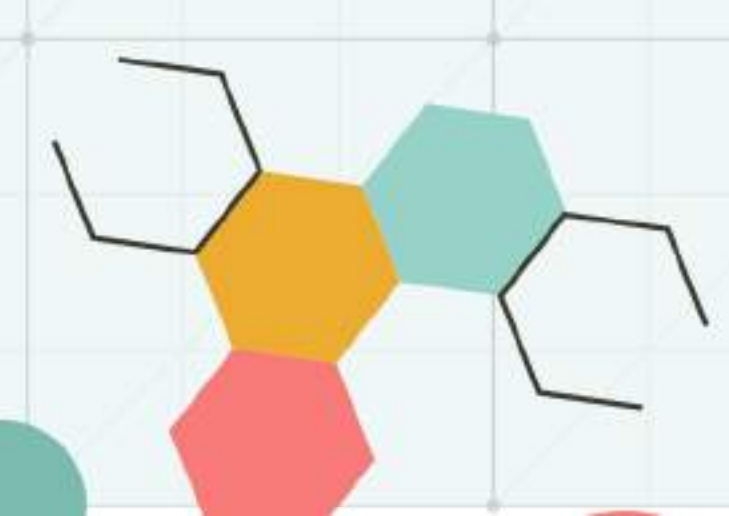
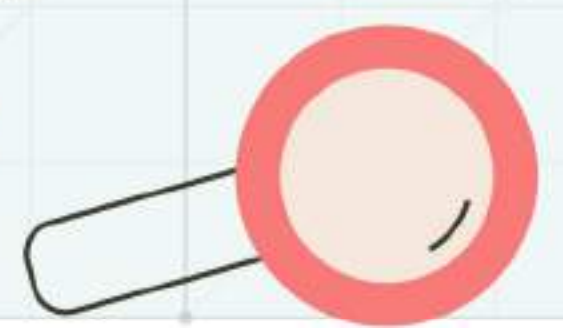
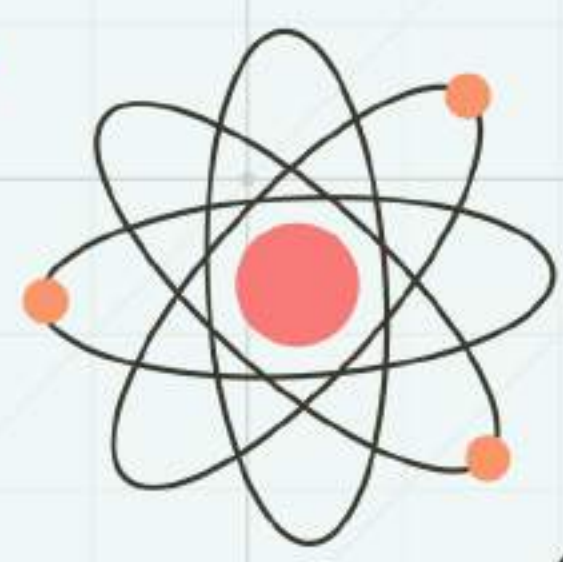
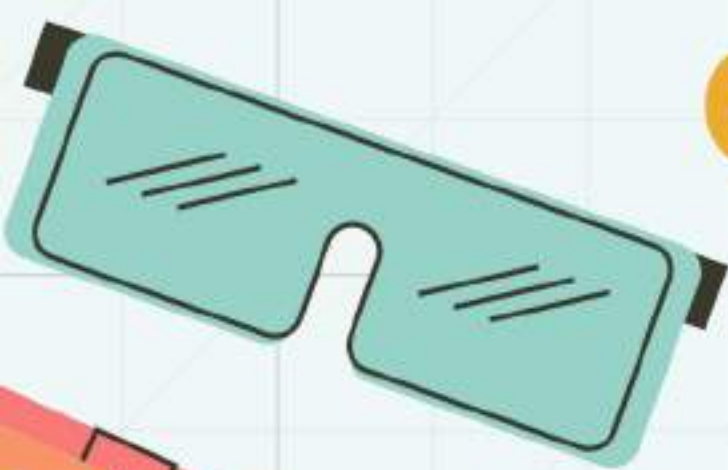
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Rate of heat flow = $\frac{dQ}{dt}$ = Heat Current

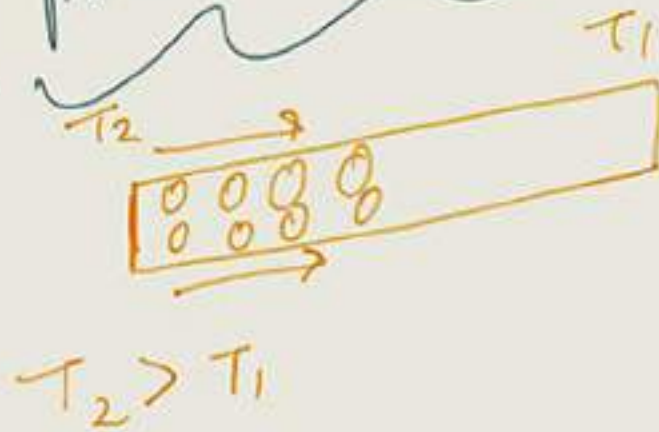
Heat transfer

model

transfer of energy due to the presence of temperature difference

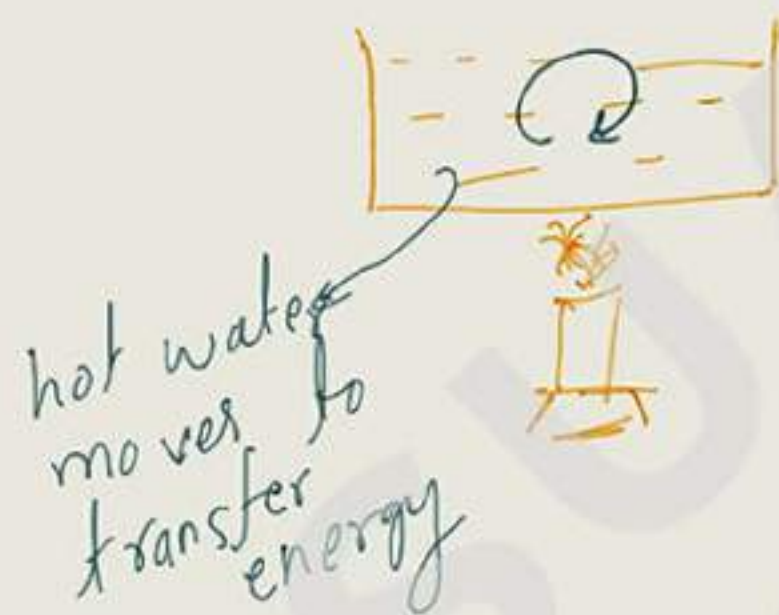
Conduction

Heat transfer through the medium via energy transfer from particle to particle



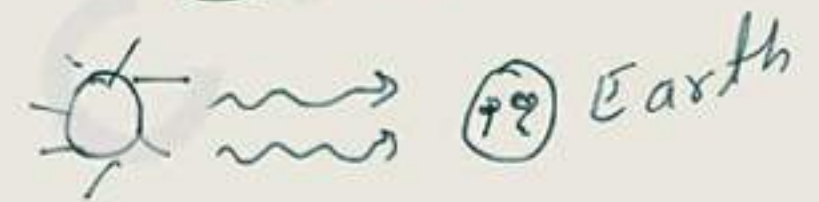
Convection

heat transfer from one part to other by motion of medium itself



Radiation

heat transfer using E-M waves. It does not need a medium for energy transfer.



Variable state: When the variables (density, temperature etc) at a particular point change with time

Steady state:

When the variables associated with a particular point do not change with time.

i.e.

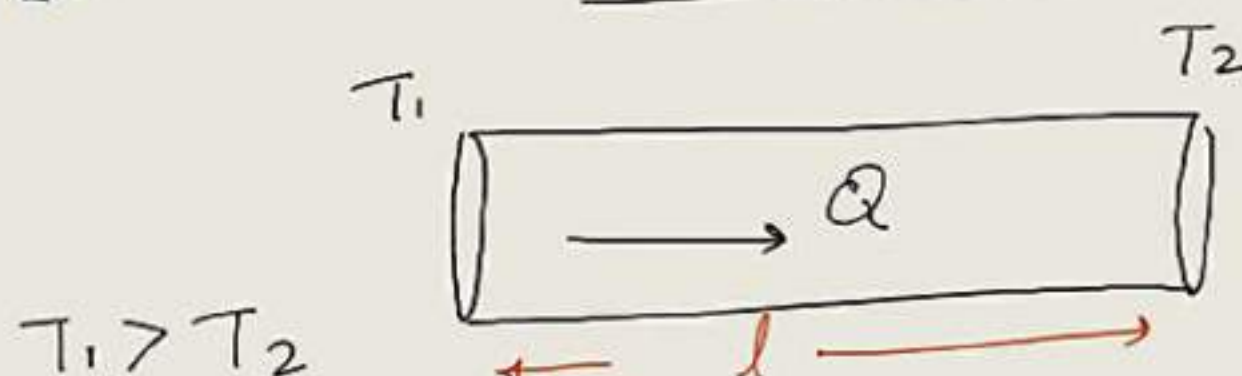
$$\left\{ \begin{array}{l} \left(\frac{\partial T}{\partial x} \right)_t \neq 0 \\ \left(\frac{\partial T}{\partial t} \right)_t \neq 0 \end{array} \right\}$$

$$\left\{ \begin{array}{l} \left(\frac{\partial T}{\partial x} \right)_t = 0 \\ \left(\frac{\partial T}{\partial t} \right)_t = 0 \end{array} \right\}$$

steady state

variable state

Conduction:



$$\frac{dQ}{dt} = \dot{Q}$$

$$\dot{Q} \propto A \quad \text{Area of cross-section}$$

$$\dot{Q} \propto (T_1 - T_2) \text{ or } \Delta T \quad \text{Temperature difference}$$

$$\dot{Q} \propto \frac{1}{l} \quad \text{length}$$

$$\dot{Q} \propto \frac{A \Delta T}{l} \rightarrow \left\{ \dot{Q} = \frac{K A \Delta T}{l} \right\}$$

$\left\{ \text{Thermal conductivity} \right\}$

Ability to conduct heat
represent!

different for different material.

$$\dot{Q} \rightarrow \text{SI unit} \rightarrow \frac{\text{W}}{\text{m} \cdot \text{C}}$$

$$\left\{ \dot{Q} = H \rightarrow \text{Heat current} \right\}$$

Analogy:-

Heat

$$\dot{Q} = H = \text{heat current}$$

$$\Delta T = \text{Temperature difference}$$

$$\left\{ H = \left(\frac{KA}{l} \right) \Delta T \right\}$$

Electricity

$$\frac{dq}{dt} = I = \text{Current}$$

$$\Delta V = \text{potential difference}$$

$$\left\{ I = \frac{\Delta V}{R} \right\}$$

So we can say

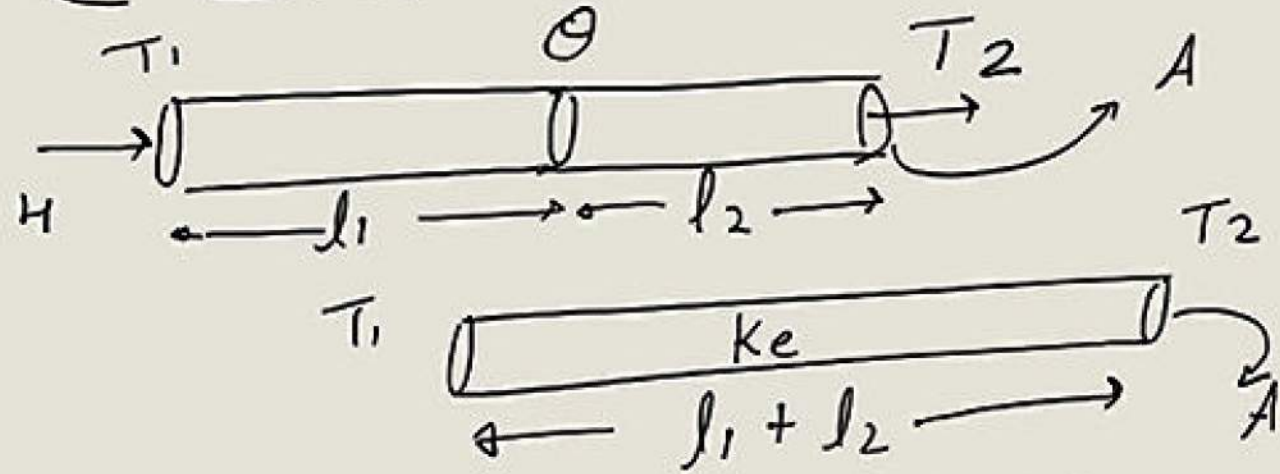
$$\Rightarrow R_{TH} = \frac{l}{KA}$$

Thermal res is like

$$\left\{ \begin{array}{l} \Delta V \rightarrow \Delta T \\ I \rightarrow H \\ R \rightarrow \frac{l}{KA} = R_{TH} \end{array} \right\}$$

{ Combination of conductors : }

{ Series combination }



$$H = \frac{T_1 - \theta}{R_1}, \quad H = \frac{\theta - T_2}{R_2}$$

$$T_1 - \theta = HR_1, \quad \theta - T_2 = HR_2$$

$$T_1 - T_2 = H R_{eq}$$

$$R_{eq} = R_1 + R_2$$

$$\frac{l_1 + l_2}{k_e} = \frac{l_1}{k_1} + \frac{l_2}{k_2}$$

{ Radiation }

{ Prevost's theory }

All body emits radiation at all temperature

All body absorbs radiation at all temperature

Amount of emission/absorption depends on temperature and material

$$\{ T_{surrounding} > T_{body} \}$$

Absorption > Emission

Body temperature ↓

$$\{ T_{surrounding} < T_{body} \}$$

Absorption < Emission

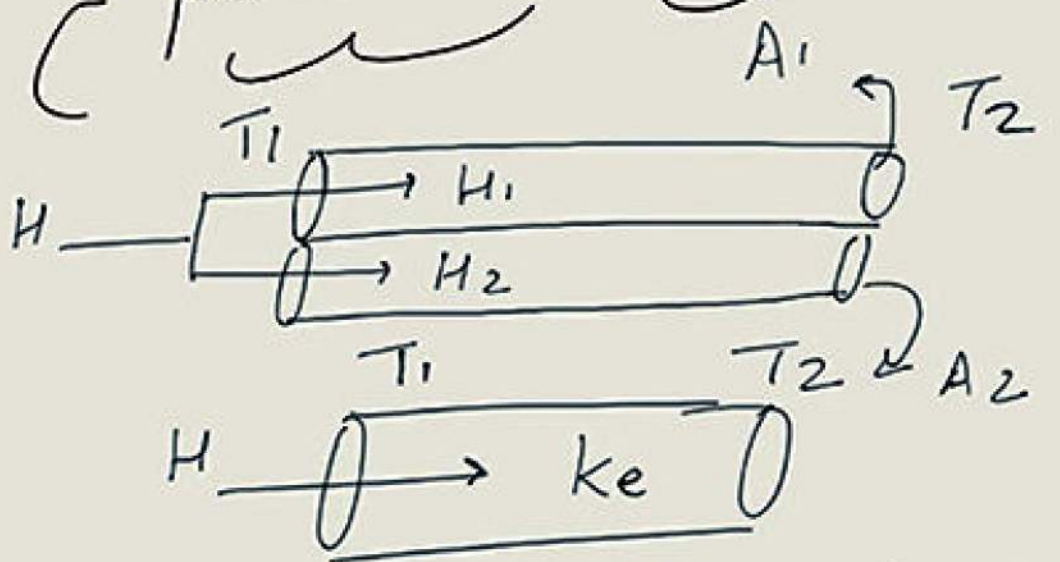
body temperature ↑

Body temperature constant

$$\{ T_{surrounding} = T_{body} \}$$

$$\{ E_{mission} = A_{bsorption} \}$$

{ parallel combination }



$$H_1 = \frac{T_1 - T_2}{R_1}, \quad H_2 = \frac{T_1 - T_2}{R_2}$$

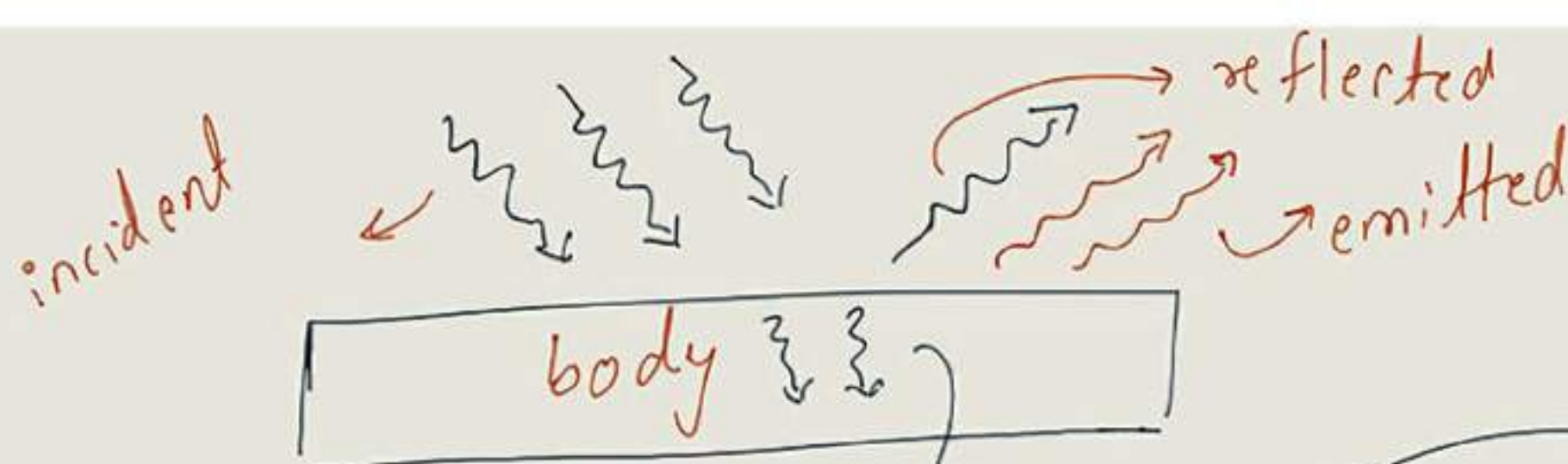
$$H = H_1 + H_2 = \frac{T_1 - T_2}{R_{eq}}$$

$$\frac{T_1 - T_2}{R_{eq}} = \frac{T_1 - T_2}{R_1} + \frac{T_1 - T_2}{R_2}$$

$$\frac{1}{R_{eq}} = \frac{1}{R_1} + \frac{1}{R_2}$$

$$k_e (A_1 + A_2) = \frac{k_1 A_1}{l_1} + \frac{k_2 A_2}{l_2}$$

$$k_e = \frac{\frac{k_1 A_1}{l_1} + \frac{k_2 A_2}{l_2}}{A_1 + A_2}$$



{ Absorptive power
(a) }

absorbed

{ Amount of heat absorbed
Amount of heat incident }

{ Reflecting power
(r) }

{ Amount of heat reflected
Amount of heat incident }

{ Transmitting power
(t) }

{ Amount of heat transmitted
Amount of heat incident }

{ $a + r + t = 1$ }

{ emissive power
(E) }

{ $\frac{Q}{At}$ }

Heat emitted at given temperature

time

Area

{ Coefficient of emission = e }

{ $\frac{E}{E_b}$ }

emissive power of
body at " T "

Temp

emissive power of
black body at " T "

{ An ideal body that absorbs
all radiation falling on it }

{ Emissive power
depends on }

Material
wavelength

Temperature

{ E_λ }

spectral
emissive power
[at given wavelength]

similarly $\{a_\lambda\}$ spectral absorptive power
[at particular wavelength of radiation]

for black body
 $\{a_\lambda = 1, e_\lambda = 1\}$

* for a given condition, black body radiates / absorbs maximum amount of radiation.

{ practical black body

Lamp black (1% reflection)
Graphite (3% reflection)
platinum

isothermal enclosure



{ Kirchhoff's law

$\left\{ \frac{E}{a} = E_b \right\}$ [for any substance]

$\{ \text{true for any wavelength} \}$

OR
 $a = \frac{E}{E_b} = e$

$\left\{ \frac{E_\lambda}{a_\lambda} = E_{\lambda b} = \text{constant} \right\}$

Wien's displacement law: (Can be used to measure star's temp)

$\{ T_3 > T_2 > T_1 \}$

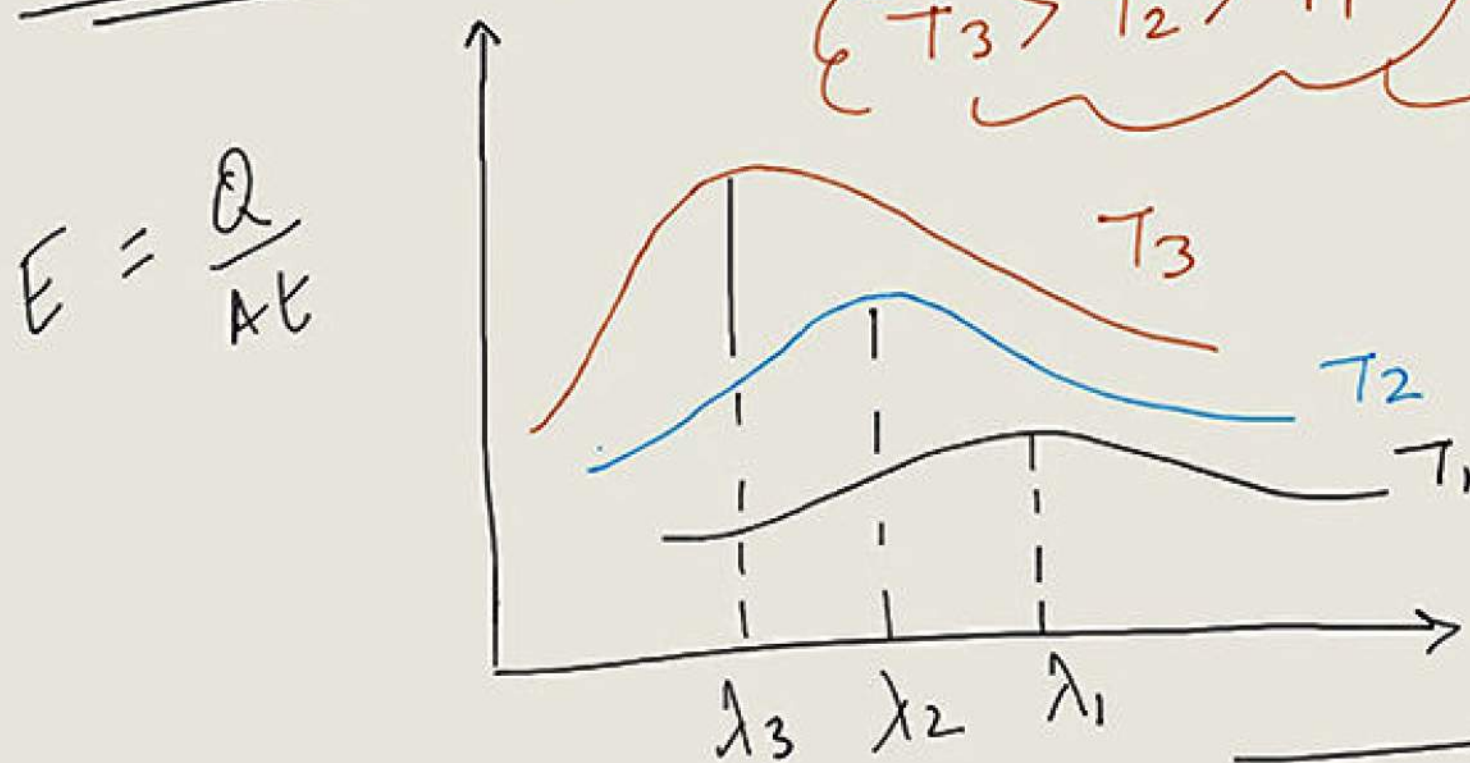
$\{ \lambda_3 < \lambda_2 < \lambda_1 \}$

$\{ \lambda_m T = \text{constant} = b \}$

Wien's displacement law

for black body

$\{ b = 2.898 \times 10^{-3} \text{ m}\cdot\text{K} \}$



* $\{ T_3 > T_2 > T_1 \}$ $\rightarrow$ $\{ I_3 > I_2 > I_1 \}$
 $\rightarrow$ $\{ \lambda_3 < \lambda_2 < \lambda_1 \}$
 Intensity of radiation

Stefan's Boltzmann law:

for perfectly black body

$\{ I \propto T^4 \}$
 $\rightarrow$ $\{ \dot{Q} = \sigma A T^4 \}$
 (power radiated) $\rightarrow$ absolute Temperature (K)
 (rate of heat/radiation emission) $\rightarrow$ Area
 Stefan's Boltzmann constant = universal constant
 $= 5.67 \times 10^{-8} \text{ W/m}^2 \text{K}^4$

for other material

$\{ 0 \leq e \leq 1 \}$
 $\{ \dot{Q} = e \sigma A T^4 \}$
 (power emitted)
 $\{ \text{when surrounding is at } T_0 \}$
 $\{ \dot{Q} = \text{net heat emitted} = e \sigma A (T^4 - T_0^4) \}$
 (power absorbed)
 $\{ a \sigma A T_0^4 = e \sigma A T_0^4 \}$
 $\therefore e = a$

if temperature difference between body and surrounding is less

Newton's law of cooling

$$T - T_0 = \Delta T$$

small

T = body's temperature
 T_0 = surrounding temperature

Rate of cooling $\propto \Delta T$

rate of change of body temperature $\propto \Delta T$

$$\frac{dT}{dt} \propto \Delta T$$

$$\frac{dT}{dt} \propto (T - T_0)$$

or
$$\frac{dT}{dt} = k(T - T_0)$$

proportionality constant depends on material & surrounding condition

derivation of Newton's law of cooling

$$\dot{Q} = e \sigma A (T^4 - T_0^4)$$

$$mc \frac{dT}{dt} = e \sigma A (T^4 - T_0^4)$$

$$\frac{dT}{dt} = \left(\frac{e \sigma A}{mc} \right) (T^4 - T_0^4)$$

$$T = T_0 + \Delta T$$

$$\Rightarrow \frac{dT}{dt} = \left(\frac{e \sigma A}{mc} \right) [(T_0 + \Delta T)^4 - T_0^4] = \left(\frac{e \sigma A T_0^4}{mc} \right) \left[\left(1 + \frac{\Delta T}{T_0} \right)^4 - 1 \right]$$

$$\frac{dT}{dt} = \left(\frac{e \sigma A T_0^4}{mc} \right) \left[\left(1 + 4 \frac{\Delta T}{T_0} \right) - 1 \right] \text{ if } \left(\frac{\Delta T}{T_0} \ll 1 \right)$$

$$\frac{dT}{dt} = \left(\frac{4 e \sigma A T_0^4}{mc} \right) (\Delta T) = k [T - T_0]$$

$$k = \frac{4 e \sigma A T_0^4}{mc}$$